Dog No	Name	23U344
Reg. No		

# B.Sc. DEGREE END SEMESTER EXAMINATION : NOVEMBER 2023 SEMESTER 3 : PHYSICS

COURSE: 19U3CPCHE3.1: ADVANCED PHYSICAL CHEMISTRY - 1

(For Regular - 2022 Admission and Improvement/Supplementary - 2021/2020/2019 Admissions)

Time : Three Hours Max. Marks: 60

#### PART A Answer All (1 mark each)

- 1. Give an example for a tetragonal crystal.
- 2. Give the eutectic composition of lead-silver system.
- 3. Which physical phenomenon is used in gas masks?
- 4. Write down the number of atoms associated in the unit cell of an element with simple cubic structure.
- 5. Define an improper rotation axis.
- 6. How many phases are present in a homogeneous system?
- 7. Who discovered carbon nanotubes?
- 8. The liquid crystals possessing limiting mobility and flowing in layers are of ...... type.

  (1 x 8 = 8)

### PART B Answer any 6 (2 marks each)

- 9. State phase rule and define the terms involved.
- 10. Suggest a green method to synthesize nanogold. Explain the procedure.
- 11. Represent different axes of rotation in benzene and identify the principal axis.
- 12. Draw the Miller indices of (100) and (111) Plane.
- 13. Discuss the effect of temperature on adsorption.
- 14. Why does ZnO appear yellow on heating?
- 15. Briefly explain the intermolecular forces present in liquids.
- 16. Distinguish between Schottky defect and Frenkel defect.

 $(2 \times 6 = 12)$ 

## PART C Answer any 4 (5 marks each)

- 17. Explain how the phase rule is modified for applying to a condensed system. Draw a general phase diagram for a simple euctectic system A-B.
- 18. An element crystallises with a FCC structure with a unit cell edge of 388 pm. Its density is 12.16 gcm<sup>-3</sup>. Calculate the atomic mass of the element.
- 19. Differentiate between chemical adsorption and physical adsorption?
- 20. What are Nanocomposites? Discuss their properties and applications.
- 21. Compare smectic, nematic and cholesteric liquid crystals.
- 22. Systematically predict the point group of  $C_2H_2$  and HCl.

 $(5 \times 4 = 20)$ 

#### PART D Answer any 2 (10 marks each)

- 23. a) What are the separations of the planes with Miller indices (111), (211) and (100) in a crystal in which the cubic unit cell is of side 432 pm?
  - b) The angle of a first order X-ray diffraction from a set of crystal planes with an interplanar spacing of 99.3 pm is 20.85°. What is the wavelength of the X-ray radiation?
- 24. Systematically work out the point group of the following molecules.
  - a) PCl<sub>5</sub> b) Nitrate ion c) CO<sub>2</sub> d) Benzene
- 25. a) Elaborate on Nanomedicine and drug delivery applications using different nanomaterials.
  - b) Briefly write the properties and applications of nanocomposites.
- 26. Discuss the phase diagrams of water system and sulphur system.

 $(10 \times 2 = 20)$