Reg.	No	Name	20P3006

MSc DEGREE END SEMESTER EXAMINATION - OCT. 2020 : JANUARY 2021 SEMESTER 3 : CHEMISTRY

COURSE: 16P3CHET09: INORGANIC CHEMISTRY - III

(For Regular - 2019 Admission and Supplementary - 2016 / 2017 / 2018 Admissions)

Time: Three Hours Max. Marks: 75

PART A Answer any 10 (2 marks each)

- 1. What is levitation?
- 2. Explain the structure of rhenium trioxide.
- 3. Arrange the follwing metal oxides in the increasing order of their Neel's temperature. MnO, FeO, NiO, CoO
- 4. There is no significant deviations from stoichiometry for group 2 metal oxides unlike 3-d metal oxides. Why?
- 5. Materials with metal excess and metal deficiency defects are termed as n-type and p-type semiconductors respectively. Why?
- 6. Explain the terms: Burgers vector and small angle tilt boundary
- 7. What are octahedral metal clusters? Give examples.
- 8. With the help of an example, discuss the synthesis of binuclear metal cluster.
- 9. What is thermal spalling?
- 10. Discuss on the magnetic properties of KCP.
- 11. How P_4S_4 is prepared? Give the structures of its α and β forms.
- 12. Illustrate the role of Zeolites as shape selective catalyst.
- 13. Predict the stability of $[(C_5H_5)_2Fe]^+$ using MNO rule.

 $(2 \times 10 = 20)$

PART B Answer any 5 (5 marks each)

- 14. Distinguish between spinels and inverse spinels.
- 15. Describe the Kronig-Penney model.
- 16. Discuss the nonstoichiometric defects of the 3-d metal monoxides with special reference to their electronic properties
- 17. Describe experimental methods for determining the diffusion coefficient in metallic and ionic solids
- 18. Explain the basis of the different colour exhibited by gemstones.
- 19. Discuss on the classification of ceramic materials.
- 20. Write a note on heteropoly ions. What are heteropoly blues? Give an account of their uses.
- 21. How are silicones prepared? Explain its structure and bonding.

 $(5 \times 5 = 25)$

PART C Answer any 2 (15 marks each)

- 22. Discuss the BCS theory of superconductivity.
- 23. Give a brief account on a) Order disorder transition, b) Intermetallic compounds and c) Chemical vapour deposition technique.
- 24. List out some of the heterocyclic inorganic ring systems of Sulphur and phosphorous. Give their synthesis, structure and conformations. Explain the special features of bonding.
- 25. Explain in detail the magnetic, electrical and optical properties of one dimensional soids by giving suitable examples.

 $(15 \times 2 = 30)$