Reg. No:	Name:	P405

# MSc DEGREE END SEMESTER EXAMINATION MARCH 2016 SEMESTER - 4 CHEMISTRY COURSE: P4CHET13EL- ADVANCED INORGANIC CHEMISTRY

Time: Three Hours

Max. Marks: 75

#### Section A

(Answer any ten questions. 2 mark each)

- 1. What are vanishing integrals?
- 2. Describe what is the "method of descending symmetry".
- 3. Deduce the characters of the reducible representations for the d- orbital wave function in a tetrahedral field.
- 4. What happens to the carbonyl stretching frequency in the IR spectrum of CN<sup>-</sup> on coordination with metal ions?
- 5. Predict the number of lines in the EPR spectrum of  $H_2O$  group.
- 6. What is meant by phosphorescence?
- 7. Explain Moore's law and how is useful for nano materials.
- 8. What are nano sensors?
- 9. How is carbon nano tube synthesized?
- 10. Describe the basis and procedure of sampling.
- 11. Give the various methods of air pollution monitoring.
- 12. Explain how liquid ammonia solution of alkali metal ions show intense colour.
- 13. What are soft acids? Give two examples.

 $(10 \times 2 = 20)$ 

## Section B

### (Answer **any five** questions. 5 mark each)

14. Arrive at the SALC of atomic orbitals in octahedral complexes.

- 15. Discuss the selection rules for electronic d d transitions and their exceptions.
- 16. What are quantum dots? Give its applications.
- 17. Describe briefly on nano lithography and nano wires.
- 18. Compare the principles behind BOD and COD determinations. Give the methods employed.
- 19. Explain the technique of electrophoretic deposition.
- 20. Differentiate between essential and non essential water samples. How can you eliminate

moisture in a sample?

21. Explain the non-aqueous solvents like anhydrous sulphuric acid and hydrogen halides. Give its applications.

(5 x 5 = 25) (PTO)

#### Section C

#### (Answer any two questions. 15 mark each)

- 22. Construct molecular orbital energy level diagram for an octahedral complex involving  $\pi$  bonding ligands.
- 23. Explain the self-assembled mono layers on gold-growth process and its phase transitions, including formation, detection and analysis of gas phase clusters.
- 24. What is the isomer shift in Mössbauer spectroscopy? How is it used in the study of Fe(II) complexes, giving its spectrum?
- 25. Describe the use of IR and Raman spectroscopy in the structural elucidation of coordination complexes.

 $(15 \times 2 = 30)$ 

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